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Electronic structure and thermoelectric properties of bismuth telluride and bismuth selenide

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Abstract. The electronic structures of the two thermoelectric materials Bi_2Te_3 and Bi_2Se_3 are studied using density-functional theory with the spin-orbit interaction included. The electron states in the gap region and the chemical bonding can be described in terms of $\text{pp}\sigma$ interaction between the atomic p orbitals within the ‘quintuple’ layer. For Bi_2Se_3 , we find both the valence-band maximum as well as the conduction-band minimum, each with a nearly isotropic effective mass, to occur at the zone centre in agreement with experimental results. For Bi_2Te_3 , we find that the six valleys for the valence-band maximum are located in the mirror planes of the Brillouin zone and they have a highly anisotropic effective mass, leading to an agreement between the de Haas-van Alphen data for the p-doped samples and the calculated Fermi surface. The calculated conduction band, however, has only two minima, instead of the six minima indicated from earlier experiments. The calculated Seebeck coefficients for both p-type and n-type materials are in agreement with the experiments.

Bismuth telluride and its selenide alloys are the classic thermoelectric materials with large Seebeck coefficients and, as such, they have been used in thermoelectric refrigeration for a long time [1, 2]. There have been several attempts to fabricate better thermoelectric materials, and materials such as skutterdite pnictides and superlattices consisting of bismuth-telluride-type layers have been recently proposed [3–7]. In spite of their technological importance and the wealth of existing experimental data on Bi_2Te_3 and Bi_2Se_3 , surprisingly enough, serious studies of the electronic structure of these compounds based on theoretical calculations have been started only recently.

Several early attempts [8–10] at the theoretical determination of the electronic band structure of Bi_2Te_3 relied on the empirical pseudopotential approach. To our knowledge, the only density-functional calculation for Bi_2Te_3 reported in the literature was performed by Thomas *et al* [11] using the LAPW method. However, in their calculation, even though the spin-orbit coupling effects were recognized to be large and significant, the spin-orbit coupling was included only empirically with a tight-binding fitting procedure, which was not adequate for describing the important details of the valence- and conduction-band valleys of these materials, which are small-band-gap semiconductors. The only band-structure calculation of the other compound, namely Bi_2Se_3 , reported in the literature is an early calculation using a pseudopotential technique [12]. In this paper, we study the electronic band structures of Bi_2Te_3 and Bi_2Se_3 on the basis of density-functional calculations taking spin-orbit coupling effects into account.

The crystal structures of both Bi_2Se_3 and Bi_2Te_3 are rhombohedral with the space group D_{3d}^5 ($R\bar{3}m$) [13, 14] with five atoms in the trigonal unit cell. The structure is most simply

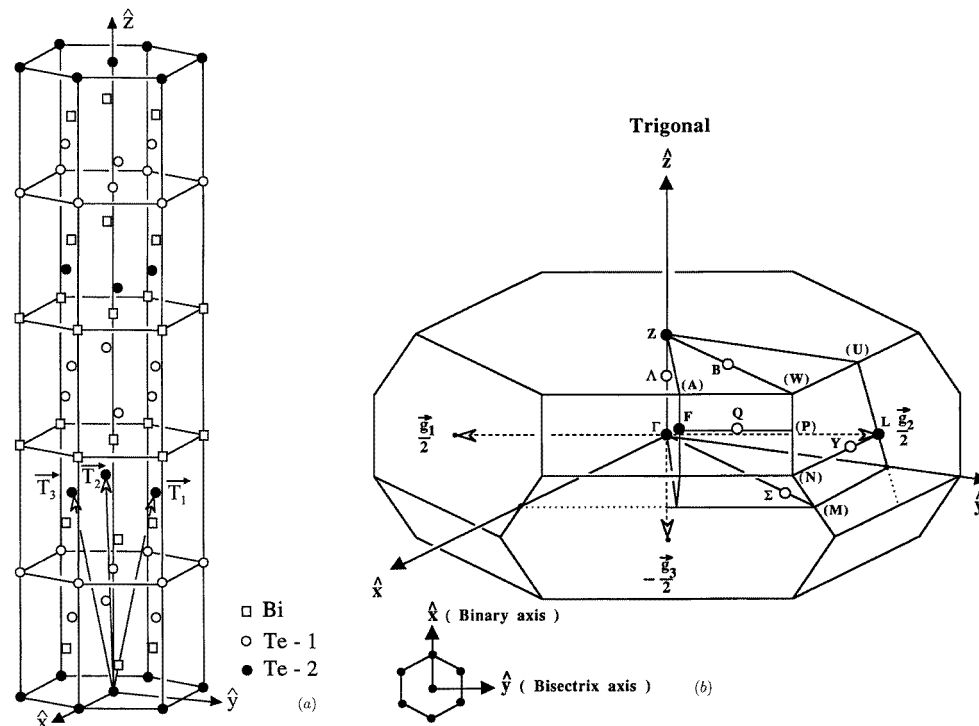


Figure 1. Crystal structures of Bi₂Se₃ and Bi₂Te₃ (a). The open and dark circles denote the Te and the Se atoms, with the Bi atoms shown as squares. The irreducible BZ is shown in (b). The points in the BZ shown in parentheses are strictly not points of symmetry. The binary axis (with twofold rotation symmetry) is along x , the bisectrix axis (occurring in the reflection plane) is along y , and the trigonal axis (with threefold rotation symmetry) is along z .

Table 1. Atom positions and sphere sizes for the LMTO calculations for Bi₂Te₃ and Bi₂Se₃. E denotes an empty sphere.

Atom	Site index	Muffin-tin sphere radius (Å)	
		Bi ₂ Te ₃	Bi ₂ Se ₃
Bi	2c($x = 0.399$)	1.78	1.75
Te1/Se1	2c($x = 0.792$)	1.76	1.58
Te2/Se2	1a	1.88	1.69
E1	1b	1.45	1.44
E2	2c($x = 0.10$)	1.28	1.22
E3	2c($x = 0.30$)	1.26	1.18
Space group:	D _{3d} ⁵ ($R\bar{3}m$), rhombohedral unit cell		
Lattice constant, a :		10.45 Å	9.84 Å
Rhombohedral angle, α :		24°8'	24°8'

visualized in terms of a layer structure and a hexagonal cell sketched in figure 1. The structure may be described as quintuple layers of atoms ('quintuple layer leaves'), which are stacked along the c -axis. The five individual atomic layers making up the quintuple layer occur in the sequence Te(1)–Bi–Te(2)–Bi–Te(1), where Te(1) and Te(2) denote the two

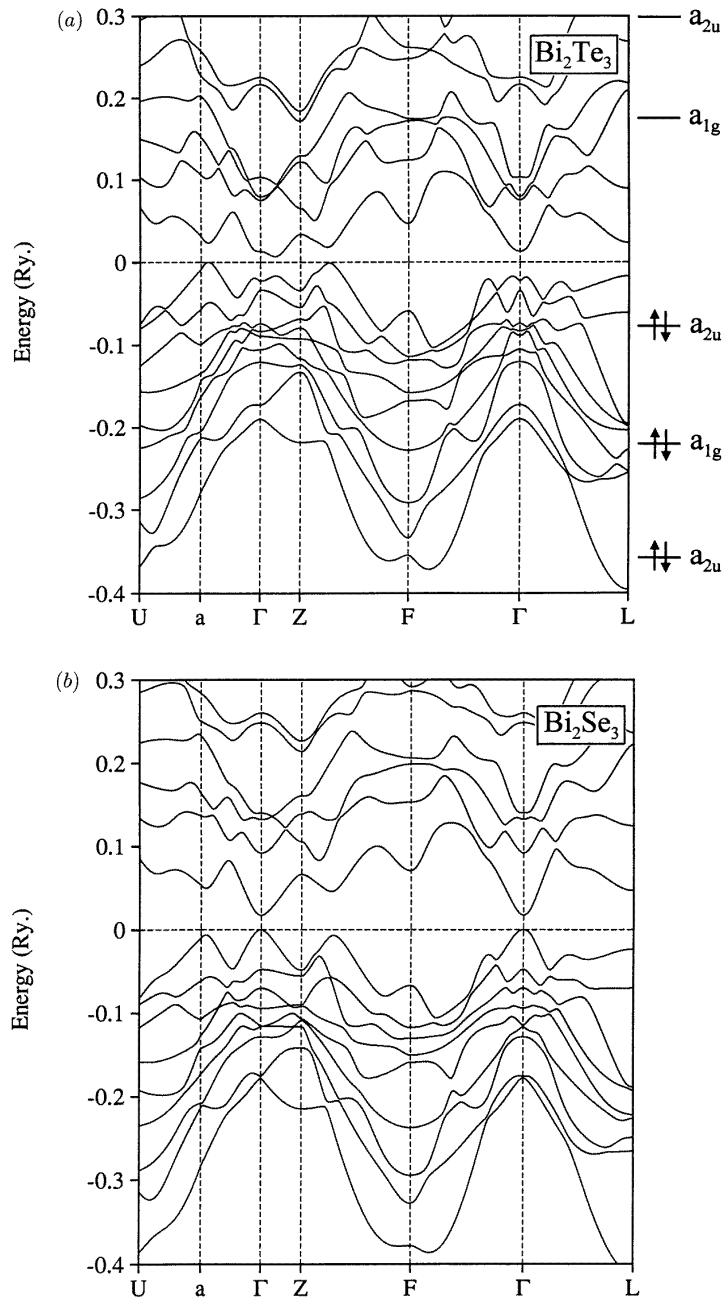


Figure 2. Electronic band structures for Bi_2Te_3 (a) and Bi_2Se_3 (b) obtained from LMTO-LDA calculations with spin-orbit coupling included. For Bi_2Te_3 , the valence-band maximum occurs along the Γ -a direction (the inset of figure 3 shows the location of the point 'a' in the BZ). The valence-band top along the Z-F direction has energy comparable to the valence-band maximum because these two k -points lie close by in the BZ. For Bi_2Te_3 , the energies of the five molecular orbitals (MOs) obtained from a simple 5×5 tight-binding model are indicated. Each of the MOs produces three bands because of the threefold degeneracy of the p orbitals.

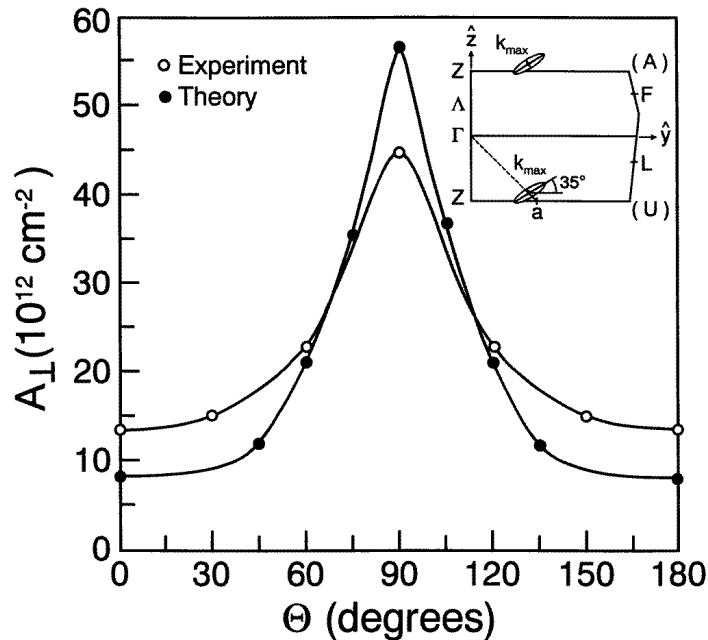


Figure 3. A comparison of the calculated extremal cross-sectional areas A_{\perp} as a function of the azimuthal angle θ with the dHvA measurement for p-doped Bi_2Te_3 with a carrier concentration of $4 \times 10^{18} \text{ cm}^{-3}$ corresponding to a hole pocket with $E_F = 21.8 \text{ meV}$. The inset shows the location of the valence-band maximum in the BZ obtained from the present calculation. Solid lines through the point k_{max} corresponding to the valence-band maximum indicate the directions of the principal axes of the mass tensor.

different types of tellurium atom in the crystal. The description in terms of the stacking of the quintuple layers is justified in view of the appreciable distances between adjacent Te(1) layers and it is an important structural feature for the description of the chemical bonding as discussed below. Each atom is surrounded by six atoms, three in the layer below and three in the layer above, along the c -axis in nearly octahedral coordination, except for the Te(1) atoms whose three nearest-neighbour Te(1) atoms, out of a total of six, are located on a different ‘quintuple’ layer leaf, and hence are somewhat further away from the octahedral coordination.

The band structures are calculated using the local-density approximation (LDA) to the density-functional theory with the linear muffin-tin orbital method in the atomic-sphere approximation (LMTO-ASA) [15]. The von Barth–Hedin exchange–correlation potential was used [16]. Muffin-tin orbitals of angular momenta s , p , and d are retained on all atoms and on the several empty spheres which were introduced in the interstitial region. Tests to see whether the inclusion of orbitals with higher angular momenta (e.g., s , p , d and f on all spheres) altered the states near the gap region were performed, but they did not affect the results significantly. Core electrons were treated self-consistently (‘soft’ cores) and with relativistic effects taken into account. The positions of the atoms and the empty spheres as well as the sphere radii used in the calculations are given in table 1.

The electron band structures for the two compounds are shown in figure 2. We find that the spin–orbit coupling introduces significant changes in the band structure for Bi_2Te_3 in

the gap region (which is consistent with the results of Thomas *et al* [11]), but it has very little effect for Bi_2Se_3 . In fact, for the former, without the spin-orbit coupling effects, one obtains a single-valley valence-band maximum and the same for the conduction-band minimum, both occurring at the Γ point.

The calculated electronic band structure for Bi_2Se_3 shows a one-valley conduction-band minimum and a one-valley valence-band maximum occurring at the centre of the Brillouin zone (BZ) and we find both valleys to have an isotropic effective mass. The nature of the conduction-band minimum has been studied experimentally [20] and it is consistent with our results. The experimental data on the nature of the valence-band maximum are, however, limited because of the difficulty of making the p-doped samples. On the basis of an early galvanomagnetic measurement, a single-valley valence-band maximum located at the Γ point was postulated [21]. This is confirmed by the present calculations. The calculated gap value is about 0.24 eV, consistent with the measured value of 0.2–0.3 eV [22].

For Bi_2Te_3 , there are six valleys for the highest valence band located in the mirror planes of the BZ at the point shown in the inset of figure 3, which shows the calculated location of the valence-band maximum in the BZ. The valleys are from experiments known to be located on the plane containing the trigonal and the bisectrix axis, which is consistent with our results. The effective masses are highly anisotropic. From symmetry considerations, the energy near the valence-band maximum can be written as

$$2m_0E/\hbar^2 = \alpha_{11}k_x^2 + \alpha_{22}k_y^2 + \alpha_{33}k_z^2 + 2\alpha_{23}k_yk_z \quad (1)$$

where \mathbf{k} is the Bloch vector, m_0 is the free-electron mass, and the mass parameters α are related to the effective-mass tensor. Our fitted values of the mass parameters are: $\alpha_{11} = 109.3$, $\alpha_{22} = 5.2$, $\alpha_{33} = 6.2$ and $\alpha_{23} = 3.1$, which leads to the principal axes of the energy ellipsoid making an angle of about 35° with the crystalline axes as shown in figure 3. This is in very good agreement with the measured value of $31.5^\circ \pm 10\%$ [17]. Figure 3 shows the satisfactory agreement between the experimental and calculated extremal cross-sectional areas of one Fermi surface ellipsoid for the p-doped sample corresponding to the Fermi energy $E_F = 21.8$ meV, with the magnetic field perpendicular to the c -axis.

The conduction-band minimum occurs along the Λ (Γ -Z) line in the BZ, which results in two valleys for the conduction-band minimum with a more-or-less isotropic mass tensor. This is in disagreement with the experiment which shows six valleys located on a mirror plane of the BZ [18]. We note that the Λ line is in fact on the mirror plane of the BZ, but because we have additional symmetry for the Λ line, only two valleys occur if the minimum is located on this line. One would get six valleys if this minimum was slightly displaced from the Λ line. A small change in the electronic band structure, resulting perhaps from the inclusion of the non-spherical part of the potential (omitted in the present atomic-sphere approximation), may be sufficient to explain the observed six-valley minimum. Within the ASA, we have considered small changes in the atomic positions of the crystal, which resulted in the same valley structure. The calculated band gap is about 0.11 eV for Bi_2Te_3 as compared to the experimental value of 0.15 eV [19].

The picture of chemical bonding can be understood in the following way. First of all, we note from the partial density of states (DOS) (figure 4) that the valence and conduction bands consist of the metal and chalcogen p orbitals, while the bands arising from the valence s orbitals occur about 0.8 Ryd below the valence-band maximum, consistent with the photoemission experiments [23]. Focusing on Bi_2Te_3 , the electron states in the vicinity of the gap region consist of the Bi(p) and the Te(p) orbitals, since all other orbitals lie far away in energy [24]. As discussed already, the structure consists of quintuple layers stacked on top of one another, with the layer sequence of Te(1)–Bi–Te(2)–Bi–Te(1) within

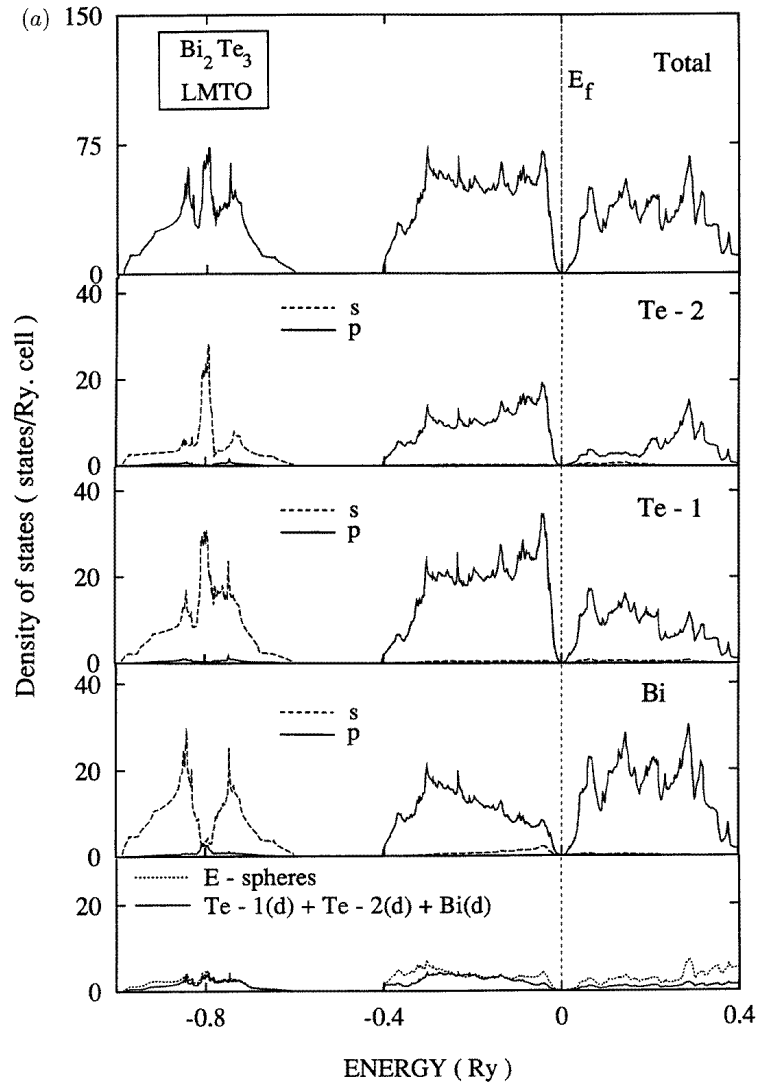


Figure 4. One-electron densities of states for (a) Bi_2Te_3 and (b) Bi_2Se_3 with spin-orbit coupling included.

each quintuple layer. Each of the three p orbitals on an atom points towards an atom on the layer above, and one on the layer below, in such a way that, since the five atoms lie roughly on a straight line, we get a sequence of five p orbitals interacting via the $pp\sigma$ interaction. (See figure 5.)

If we now keep the predominant $pp\sigma$ interaction between nearest-neighbour atoms and only within a quintuple layer, then the electron states can be described by a 5×5 tight-binding Hamiltonian matrix. There are then two parameters entering the tight-binding Hamiltonian, namely the difference of the on-site p energies and the NN $pp\sigma$ matrix element, which describes the Hamiltonian. Following Harrison [25], we take the various parameters to be: $\varepsilon_p(\text{Bi}) = -7$ eV, $\varepsilon_p(\text{Te}) = -8.5$ eV, and $V_{pp\sigma} = -2.5$ eV, which then results in the

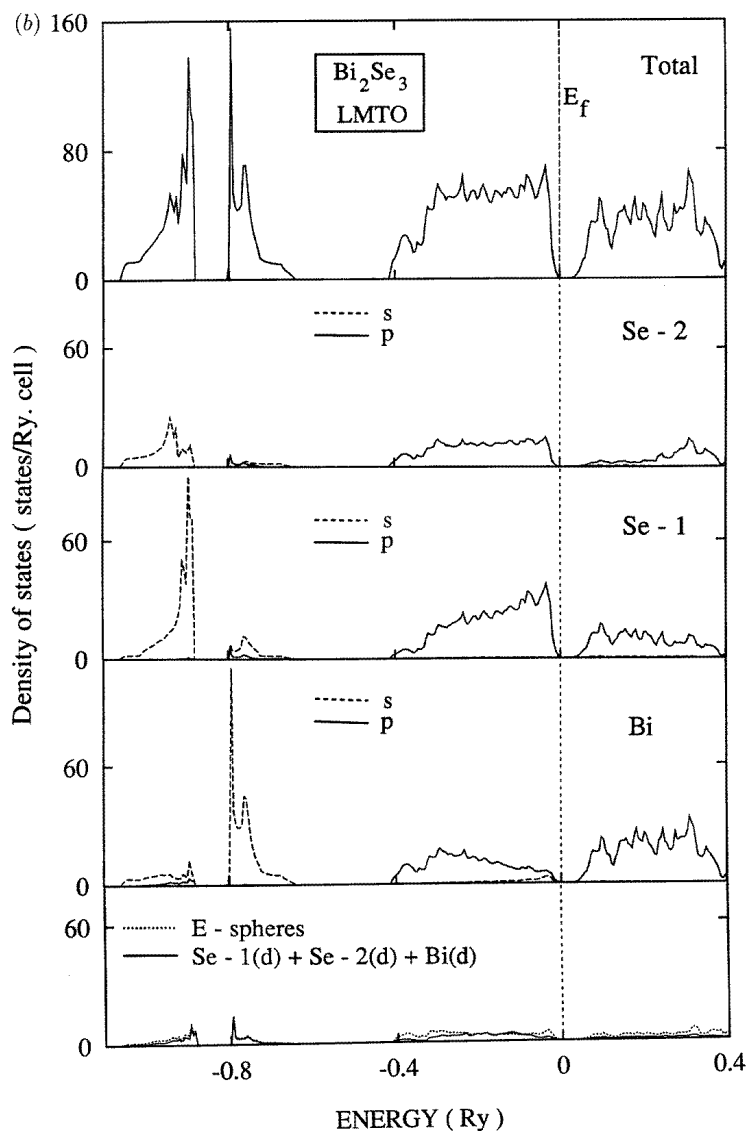


Figure 4. (Continued)

eigenvalues of $\varepsilon = -12.1, -10.4, -8.5, -5.1,$ and -3.4 eV. These are indicated along with their symmetries in figure 5 and also in the band-structure plot for Bi_2Te_3 in figure 2(a). The five p orbitals span the $2a_{1g} + 3a_{2u}$ irreducible representations of the relevant $D_{\infty h}$ point group as seen from the figure. Of these five molecular orbitals, the lowest two are bonding in character, the highest two are anti-bonding, and the remaining one is non-bonding. The valence-electron count is such that the lowest three of these orbitals are occupied, leading to the ionic charges given roughly by the chemical formula $\text{Bi}_2^{0.7+}\text{Te}(1)_2^{0.5-}\text{Te}(2)^{0.35-}$, extracted from the eigenvectors of the 5×5 Hamiltonian. (The charges do not add up to zero because of rounding.) The charge-density contour plots (figure 6) clearly show the metal atoms to be positively charged and the chalcogen atoms to be negatively charged for both Bi_2Se_3

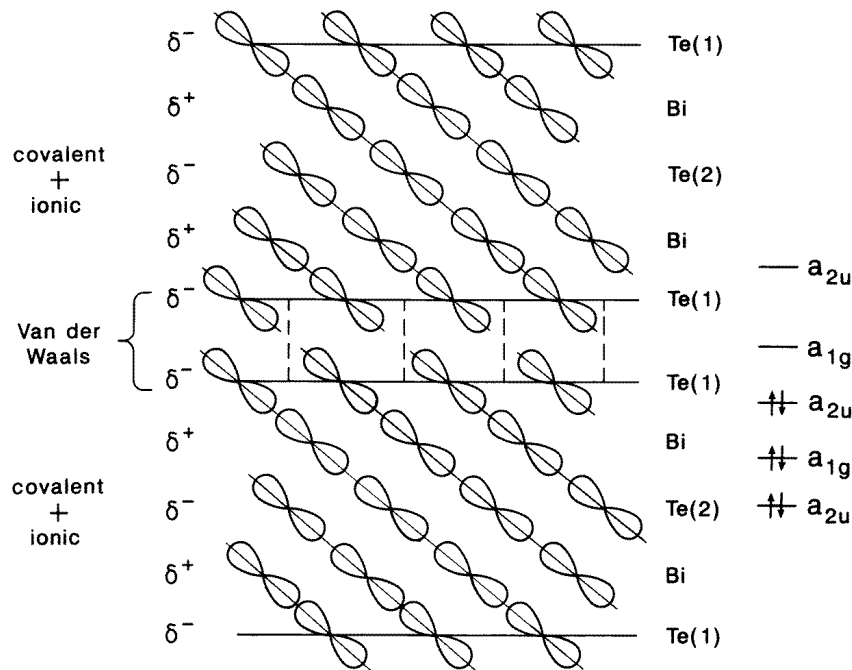


Figure 5. A sketch of the chemical bonding consistent with the results of the density-functional calculations. The molecular orbitals produced by the $pp\sigma$ interaction on the five-atom linear chain are shown on the right.

and Bi_2Te_3 providing support for the simple picture of chemical bonding. Thus the bonding between the atoms within a quintuple layer is of covalent–ionic type, while the interaction between two Te(1) layers belonging to two different quintuple layers is of the Van der Waals type as indicated in figure 5. After we completed the analysis of the chemical bonding, we became aware of the work of Pecheur and Toussaint [26], who also proposed a picture of chemical bonding governed by the $pp\sigma$ interactions similar to ours.

The thermoelectric power or the Seebeck coefficient is given by $\alpha = (2/3)C_v/ne$ from the transport equation within the approximation where the mean free path is independent of the energy of the carrier (electron or hole). (The factor of two-thirds disappears if we make the approximation that the relaxation time τ rather than the mean free path is constant [27].) For a parabolic band, the above expression simplifies to

$$\alpha = -(2/3)\pi^2 k_B^2 T / (2e\epsilon_F). \quad (2)$$

The measured value of α at room temperature is reported [1] to be $\sim 260 \mu\text{V K}^{-1}$ for p-type Bi_2Te_3 with a hole concentration of $\sim 4 \times 10^{18} \text{ cm}^{-3}$ and $\sim 250 \mu\text{V K}^{-1}$ for n-type samples with an electron concentration of $\sim 7.4 \times 10^{18} \text{ cm}^{-3}$. These compare very well with our calculated values of $313 \mu\text{V K}^{-1}$ and $196 \mu\text{V K}^{-1}$ for the p-doped and the n-doped cases, respectively, using $T = 300 \text{ K}$. Similarly, for Bi_2Se_3 , the measured value of $\sim 10 \mu\text{V K}^{-1}$ for both n-type and p-type samples with carrier concentrations of $4 \times 10^{19} \text{ cm}^{-3}$ and $3 \times 10^{19} \text{ cm}^{-3}$, respectively, at $T = 400 \text{ K}$ [28], agrees reasonably well with the calculated values of $13.5 \mu\text{V K}^{-1}$ and $12.8 \mu\text{V K}^{-1}$.

In summary, using local-density calculations we have studied the electronic structure of Bi_2Se_3 and Bi_2Te_3 and have obtained a simple picture of chemical bonding in these materials.

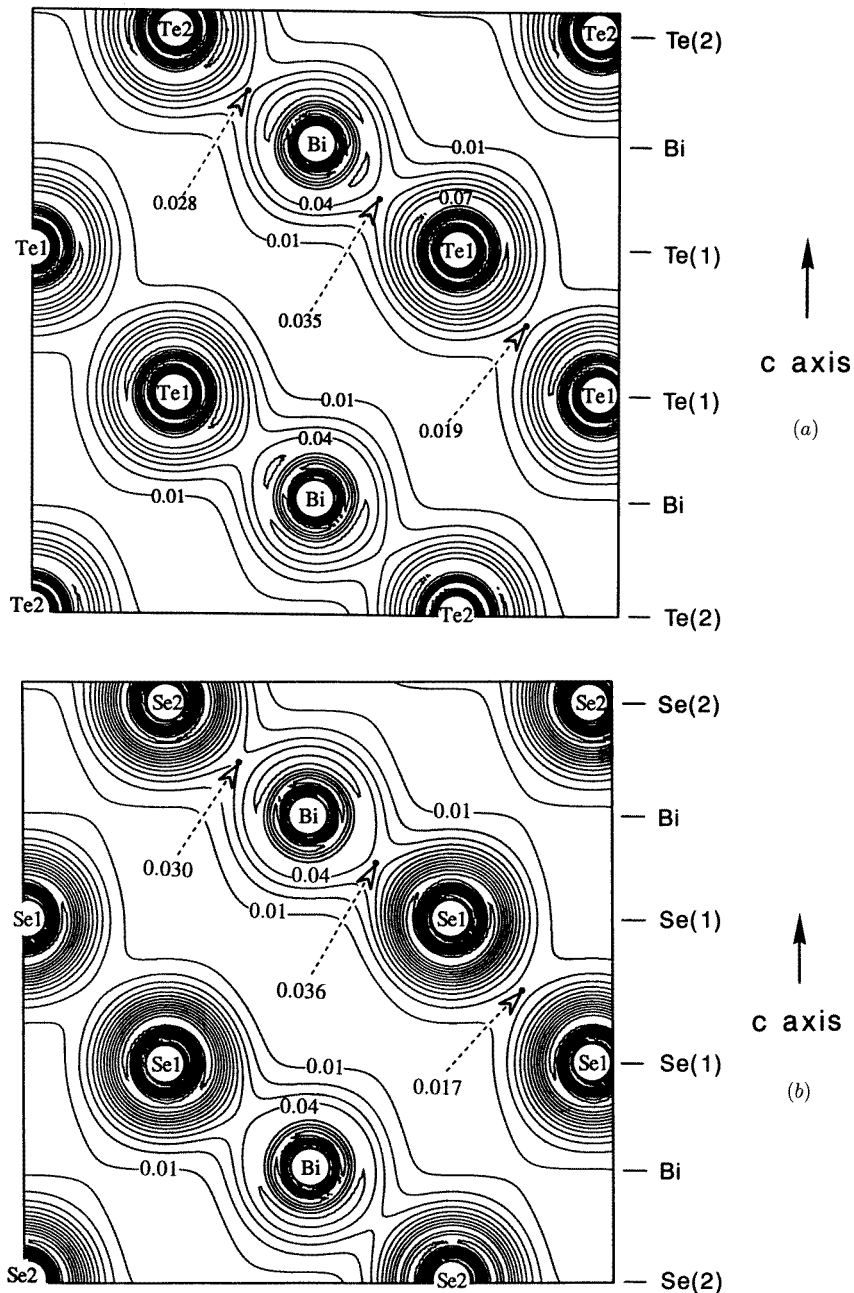


Figure 6. A valence charge-density contour plot on the xz -plane for Bi_2Te_3 (a). Contour values are in units of electrons au^{-3} . The stacking of the various layers of atoms along the c -axis is indicated on the right. The plot supports the covalent-ionic picture of chemical bonding with the metal (Bi) atoms positively charged and the chalcogen (Te or Se) atoms negatively charged. The Te(1) atoms, which form the outer boundary layers of the quintuple layer structure, are weakly bonded via van der Waals forces, leading to an appreciably smaller electron charge density at the centre of the Te(1)–Te(1) bonds. The same picture of chemical bonding is supported by the charge-density contour plot for Bi_2Se_3 (b).

The calculated thermoelectric power agrees reasonably well with the experiments for both materials.

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